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Question.1. What is
chemical kinetics ?

Answer. Chemical
kinetics is that
branch of
chemistry which
deals with the
study of the rates
of reactions and

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their mechanisms.

Question.2. What do you understand by the rate of reaction ? Answer.

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To carry out the following reactions and classify them

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as physical or
chemical changes.
a. Iron with copper
sulphate solution in
water. b. Burning
of magnesium in
air. c. Zinc with
dilute sulphuric
acid d. Heating of
copper sulphate e.
Sodium

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a scientific
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structure,
applications of the
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and compounds.
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field of organic

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(i) Whether the

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reaction is fast, slow or instantaneous i.e. the rate of the reaction is not shown by a chemical equation.

(ii) Whether the reaction goes to completion or is stopped in between i.e., the extent to which a reaction takes place is not

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known from a
chemical equation.
Q.11. Balance the
following
equations:

Previous Year Questions with Solutions - Chemical ...

Question 1. Write
the balanced
chemical equation
with the state

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Symbols of the
following reaction:
Solutions of Barium
chloride and

Sodium sulphate in
water react to give
insoluble Barium
sulphate and the
solution of Sodium
chloride. (2017D)

answer: $\text{BaCl}_2 (\text{aq})$
 $+ \text{Na}_2 \text{SO}_4 (\text{aq}) \rightarrow$
 $\text{BaSO}_4 (\text{s}) + 2$
 $\text{NaCl} (\text{aq})$ Question

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Answers: 1. What is
flow control? ... the
reactant processed
in the feed which
means that
reactants present
in the feed
transformed to

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Reaction
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products utilizing
the chemical
reaction and with
the assistance of
energy supplied to
the system. 3.

What is the
solvent?

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... The equilibrium
constant of
chemical reaction

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presence of
catalyst. (A)
Increases (B)
Decreases ...

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The chemical
equation showing
the reaction

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between lime
water and carbon
dioxide is as
follows: $\text{Ca(OH)}_2 + \text{CO}_2 \rightarrow \text{CaCO}_3 + \text{H}_2\text{O}$
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Resources.

Feedback. 1)

Chemical reaction that absorbs heat from the surroundings is said to be _____ and has a _____ ΔH at constant pressure.

Endothermic,
negative.

Exothermic,

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neutral.

Endothermic,
positive.

exothermic,
positive. 2)

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reaction
engineering
(reaction
engineering or
reactor
engineering) is a
specialty in
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engineering or
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chemistry dealing
with chemical

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Frequently the term relates specifically to catalytic reaction systems where either a ...

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Quora. 1. Why is heating done for Oxalic Acid , KMnO_4 titration? 2. Why is H_2SO_4 used instead of HCl ? 3. Explain the principle of Salt analysis.

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Who gave the laws of chemical combination?

Answer: Antoine Lavoisier and

Joseph L. Proust.

Question 2: Who

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coined the term
atom? Answer:
John Dalton.

Question 3: Atoms
are indivisible
particles, which
cannot be created
or destroyed in a
chemical reaction.
Who said this?

Answer: John
Dalton. Question 4:

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Question: How can
you predict how
much energy is
released in a
chemical reaction?

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1. Predict: In the warm-up activity, you observed how the reaction inside the chamber affected the temperature of the surrounding water. Based on what happens to the surrounding water, do you think heat energy (enthalpy) is absorbed in the

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reaction or
released? Explain.

**Question How
can you predict
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energy is
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Mole is the unit

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used to define the number of chemical substance present in a substance. It is the amount of substance which consists of the same number of chemical units as there are atoms in exactly 12 gram of pure carbon-12.

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VIVA NOTES (A)

Volumetric Analysis
(Titrations) Qs.1

What is Titration?

Ans. The process of adding one solution from the burette into another in the conical flask in order to determine

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its volume after the
completion of the
chemical reaction
is known as

Titration. Qs.2

What is

Neutralization?

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