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11: Chemical Reactions. A double-  
replacement reaction is a reaction in  
which the positive and negative ions  
of two ionic compounds exchange  
places to form two new compounds.

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rearranged in a chemical reaction.  
The substances that: react together  
are called the reactants

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11 PHChem 11.1 Describe Reactions  
1. chemistry 2. Describing Chemical Reactions <ul><li>On May 6, 1937, the huge airship Hindenburg erupted

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into a fireball. PHChem 11.1 Describe  
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equations as well. In any chemical  
reaction one or more substances,  
called

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Chemists keep track of chemical  
reactions by writing equations as  
well. In any chemical reaction one or  
more substances, called reactants, are  
converted into one or more new  
substances, called products. The  
general form of the equation for such  
a process looks like this. (11.1.1)  
Reactants      Products

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vitamins are a danger to children and  
all vitamins should be held from the  
reach of children.

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Reactions 323 with ChemASAP  
Chemical Equations Word equations  
adequately describe chemical  
reactions, but they are cumbersome.  
It ' s easier to use the formulas for the  
reactants and products to write  
chemical equations. A chemical  
equation is a representation of a  
chemical reaction; the formulas of the  
reactants (on

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Flashcards | Quizlet A chemical reaction shows the starting reagents and conditions undergoing a chemical transformation to produce reaction products. Complete combustion is the 100% efficient combustion of a fuel in oxygen with heat to produce, CO<sub>2</sub> and H<sub>2</sub>O only.  
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Chemical reactions are described with chemical equations, which symbolically present the starting

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materials, end products, and sometimes intermediate products and reaction conditions. Chemical reactions happen at a characteristic reaction rate at a given temperature and chemical concentration.

Typically, reaction rates increase with increasing temperature because there is more thermal energy available to reach the activation energy necessary for breaking bonds between atoms.

~~Chemical reaction~~—Wikipedia

Updated January 25, 2020. A chemical reaction is a process generally characterized by a chemical change in which the starting materials (reactants) are different from the products. Chemical reactions tend to involve the motion of electrons, leading to the formation and

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CHEMICAL REACTION is a change in  
which one or more reactants change  
into one or more products. DURING A  
CHEMICAL REACTION, the chemical  
bonds of reactants break, atoms  
rearrange to form no chemical bonds,

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therefore, new chemicals.

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Ch. 11: Chemical Reactions – MS.

~~MKRTCHYAN~~

A Chemical Reaction is a process that occurs when two or more molecules interact to form a new product (s).

Compounds that interact to produce new compounds are called reactants whereas the newly formed compounds are called products.

Chemical reactions play an integral role in different industries, customs and even in our daily life.

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